



# CO<sub>2</sub> Mineralization 101

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The Climate Center – Hybrid Carbon Dioxide Removal Strategies  
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# Q: Who am I to talk to you about CDR?

## Dr. Corey Myers, Co-Founder & Chief Scientist



Rice farming @ Kyoto



Soybeans @ Hokkaido Univ.



Soybeans @ Tsukuba NARO



Forested slope @ Hokkaido



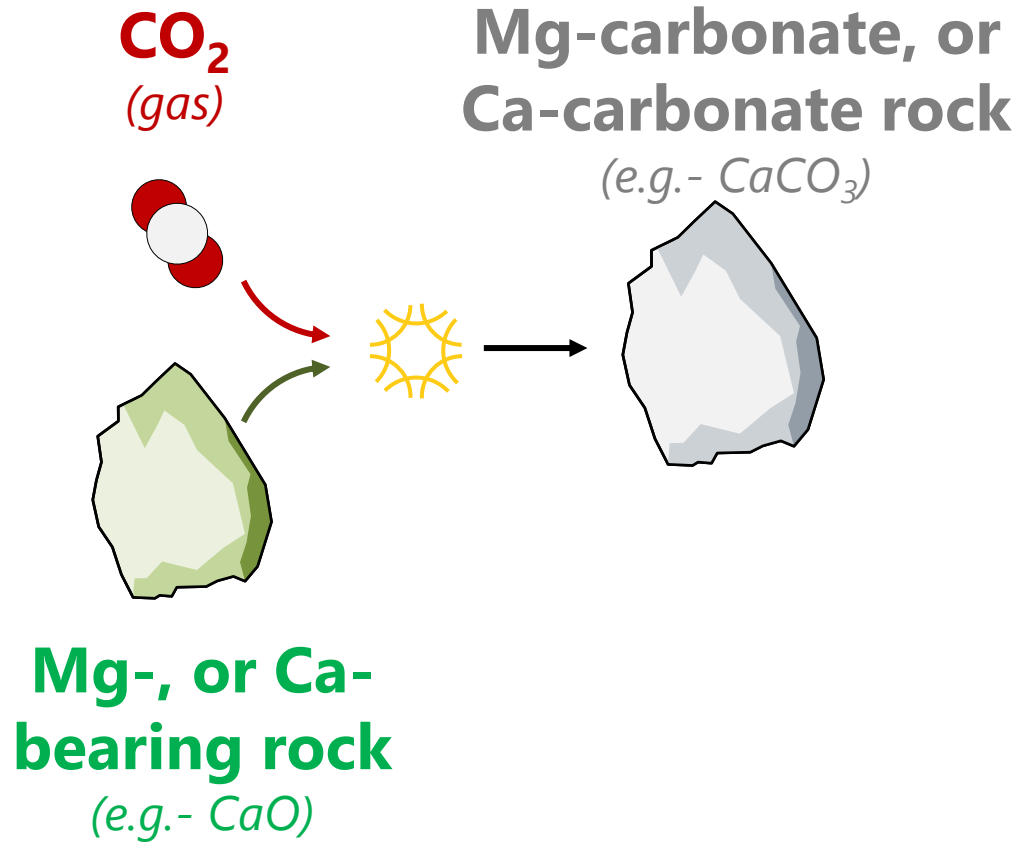
Shojin river mine entrance @ Hokkaido



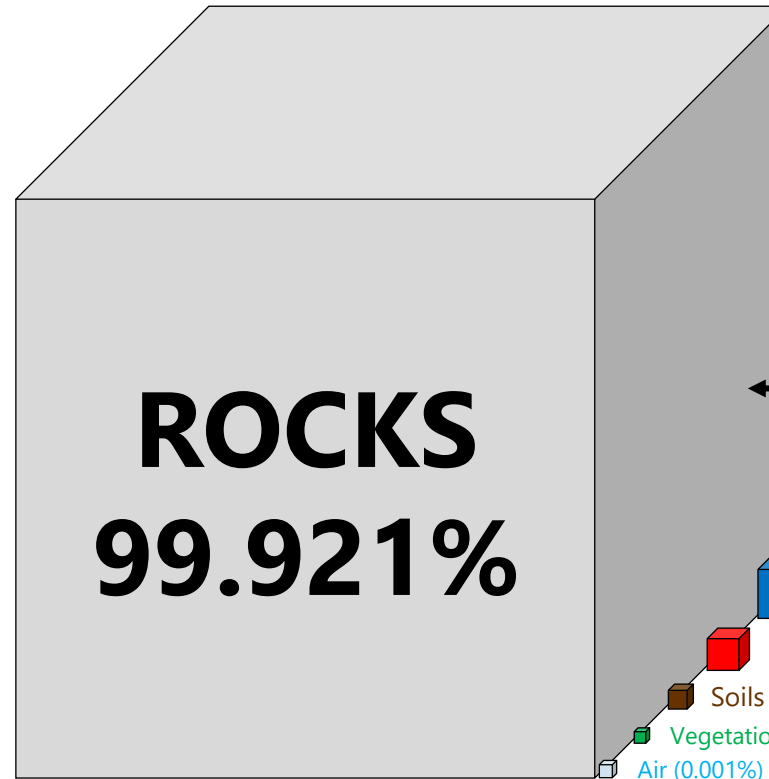
Sugarcane @ Ishigaki Island JIRCAS

- Previously, **lead engineer at Lawrence Livermore National Laboratory (LLNL)** for CO<sub>2</sub> mineralization projects, including the KCAC Carbon Mineralization Field Test—the world's largest test of surface CO<sub>2</sub> mineralization technologies.
- LLNL project lead on a DOE-funded project to **set the protocols for Monitoring, Reporting, and Verification (MRV)** for CO<sub>2</sub> mineralization-based CDR processes.
- Sat on the **CDR Innovation Roadmap team for DOE** in the areas of CO<sub>2</sub> mineralization and marine CDR.
- **Expert witness for CA state government** when considering various CDR and carbon management options.
- **Led non-pipeline CO<sub>2</sub> transport modeling for DOE.**
- Previously, Assistant Professor in Mechanical Engineering at Waseda University in Tokyo, Japan.
- A pilot plant of his CO<sub>2</sub> removal process using desalination brines is **currently being operated at a Japan's national Carbon Recycling test center** (see: <https://osakikamijima-carbon-recycling.nedo.go.jp>).
- **Japan's national Enhanced Rock Weathering** test program (A-ERW) developed directly from his work and is led by his previous lab.

# Q: What is CO<sub>2</sub> Mineralization?



A: Converting gaseous CO<sub>2</sub> into a stable, solid mineral ("rock")



Q: Is it stable  
... really?

The stores of  
carbon on Earth

induces  
climate change



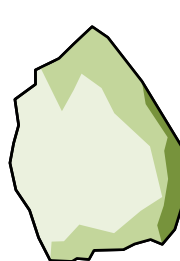
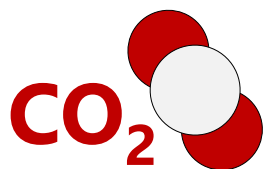
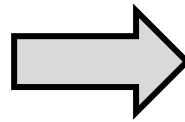
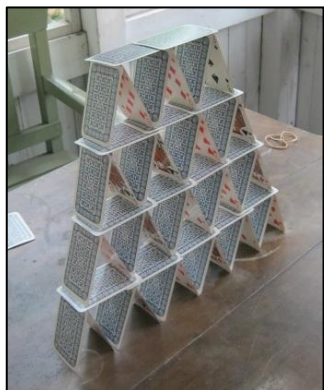
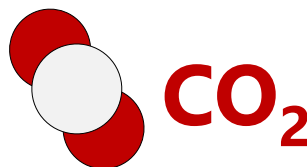
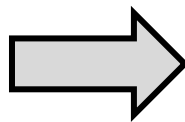
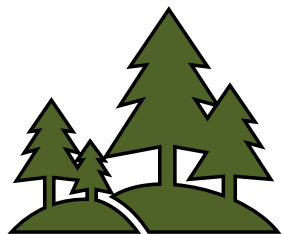
A: Yes!  
You can find it  
all around you.



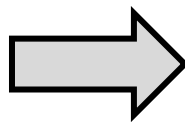
# Q: Why does this happen?

Less stable

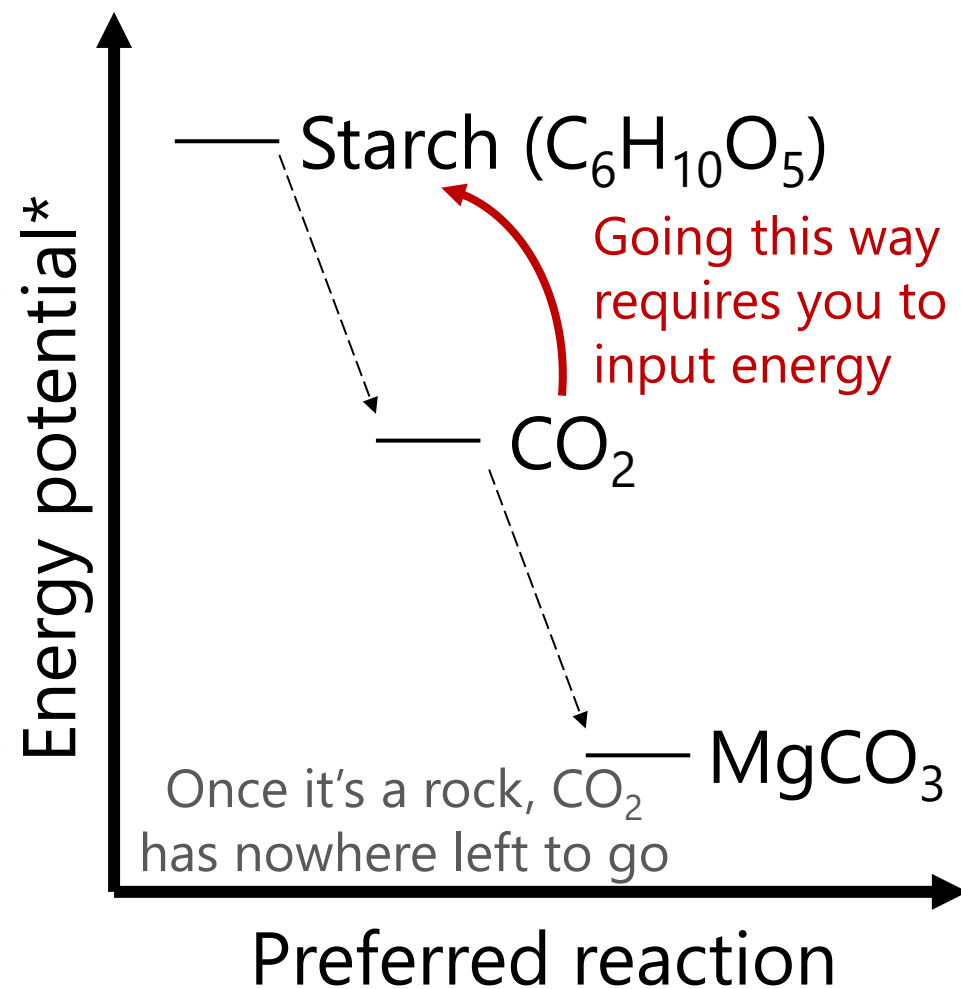
More stable



**MgO**



**MgCO<sub>3</sub>**



*\*more precisely, the change in the Gibbs Free Energy*

# Q: How do people do CO<sub>2</sub> mineralization?

Q: where do you get CO<sub>2</sub>?

Q: what do you use to bind CO<sub>2</sub>?

Q: where do you store the mineralized CO<sub>2</sub>?

Flue gas

emissions reduction (not CDR)

Air

Wastes or byproducts

emissions reduction (not CDR)

Rocks or the ocean


Below ground


Above ground

**44.01**

**Carbfix**

"in-situ"  
CO<sub>2</sub> mineralization

**Lawrence Livermore National Laboratory**



"ex-situ"  
CO<sub>2</sub> mineralization

# Q: Is this like Enhanced Rock Weathering?



A: NO.

Enhanced rock weathering (or "ERW") disperses rocks into the environment where the **rocks dissolve** (like an alka-seltzer that takes centuries to dissolve). The slightly high pH solution *might* then remove CO<sub>2</sub> from the atmosphere. **Contrast with CO<sub>2</sub> mineralization, which turns CO<sub>2</sub> into a rock.**

Whether or not CO<sub>2</sub> removal occurs is modeled but **not directly measured in ERW** (because it is too difficult and expensive to measure). **Contrast with ex-situ CO<sub>2</sub> mineralization, where the stored CO<sub>2</sub> can be held and directly measured.**

It is not known whether **by changing the base chemistry of an environment** ERW will positively or negatively impact the ecosystem. **Contrast with Anvil's process, which does not release materials into the environment.**

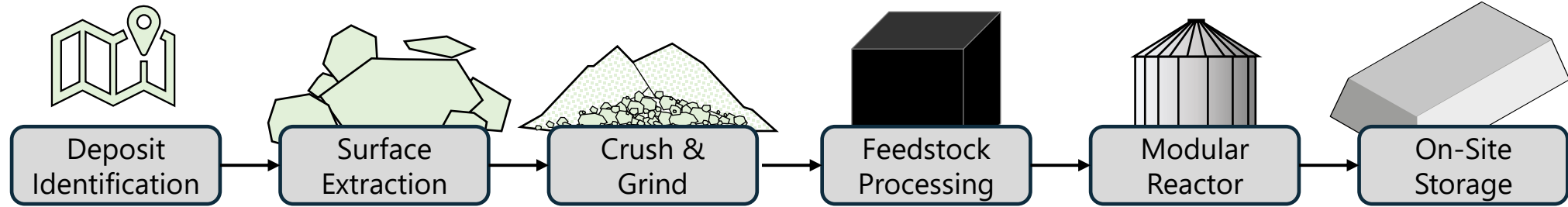
# Q: How does Anvil do CO<sub>2</sub> mineralization?



(1) Mine and grind rocks

(2) Blow air through the rocks  
(CO<sub>2</sub> becomes a rock)

(3) Put the rocks back in the ground

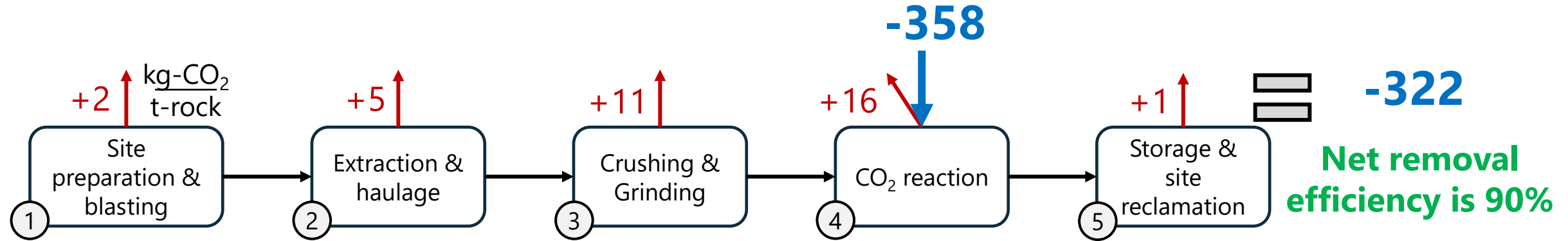


**NOTE! We don't use chemicals, high temperatures/pressures, concentrated CO<sub>2</sub>, or biomass. Just, air, water, and rock.**



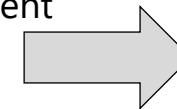


# Q: Are you removing CO<sub>2</sub> on net?



## Emissions details:

- 1) primarily loss of ecosystem carbon (4 t-CO<sub>2</sub>e/acre)<sup>a,b,c</sup> and explosives (1.5 t-CO<sub>2</sub>e/t)<sup>d</sup>
- 2) primarily diesel fuel (well-to-wheel: 12.5 kg-CO<sub>2</sub>e/gal)<sup>e,f</sup> used to operate heavy equipment
- 3) primarily electricity (Nevada: 494 g-CO<sub>2</sub>e/kWh)<sup>g</sup> used to run comminution circuit
- 4) primarily electricity (Nevada: 494 g-CO<sub>2</sub>e/kWh)<sup>g</sup> to operate reactor; no chemicals used
- 5) primarily diesel fuel (well-to-wheel: 12.5 kg-CO<sub>2</sub>e/gal)<sup>e,f</sup> used to move rocks to lined/covered pit



Using electric trucks and behind-the-meter firming, renewable energy would **increase the net CO<sub>2</sub> removal efficiency from ~90% to ~98%**



# Q: What are the benefits?

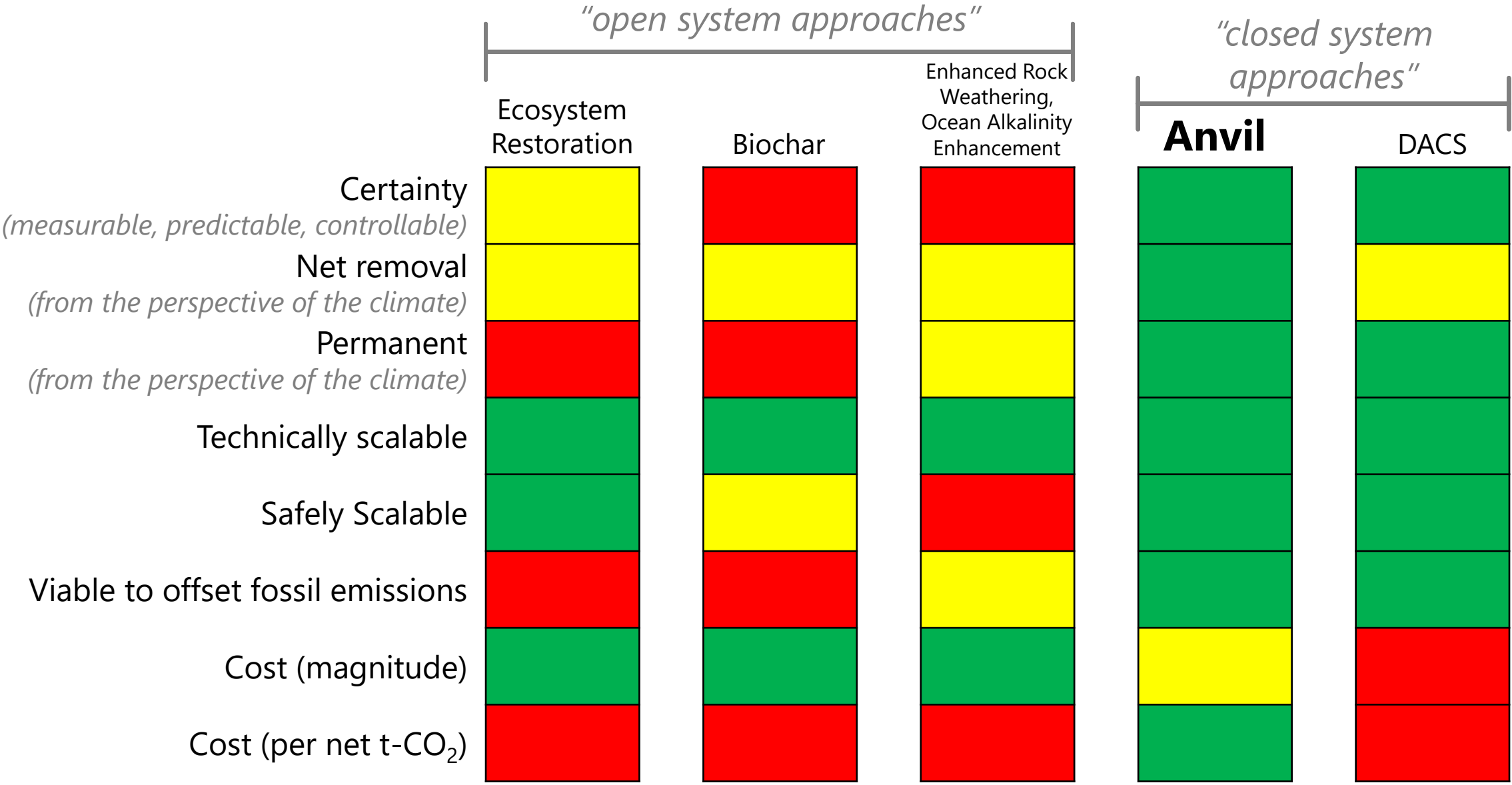


Net removal of CO<sub>2</sub> that is measurable and permanent.

Our process gives you CO<sub>2</sub> removal that is:

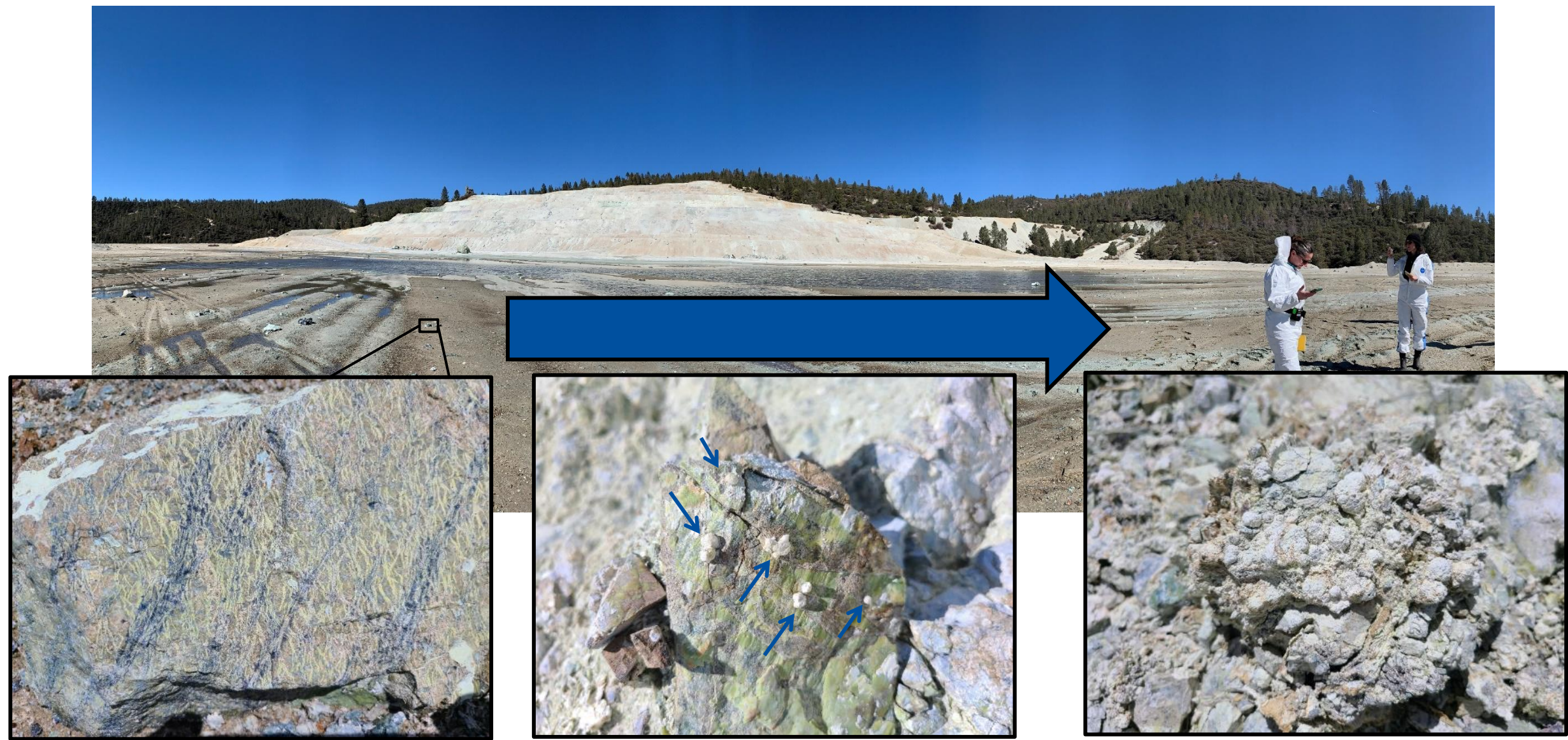
- **high removal efficiency today** (not dependent on renewables & electrification)
- **certain** (as certain as the emissions being offset)
- **auditable** (you can hold the CO<sub>2</sub> in your hand)
- **permanent** (thermodynamically stable over geologic time)
- **low energy** ( $\sim 1/10^{\text{th}}$  of direct air capture and storage)
- **a good neighbor** (no watershed impact, no truck traffic, no pipelines)
- **land sparing** (everything stays at the mine site, energy footprint << DAC)
- **low cost** (e.g., comparable to Municipal Solid Waste treatment/disposal costs)

# Q: How does Anvil compare to other CDR?



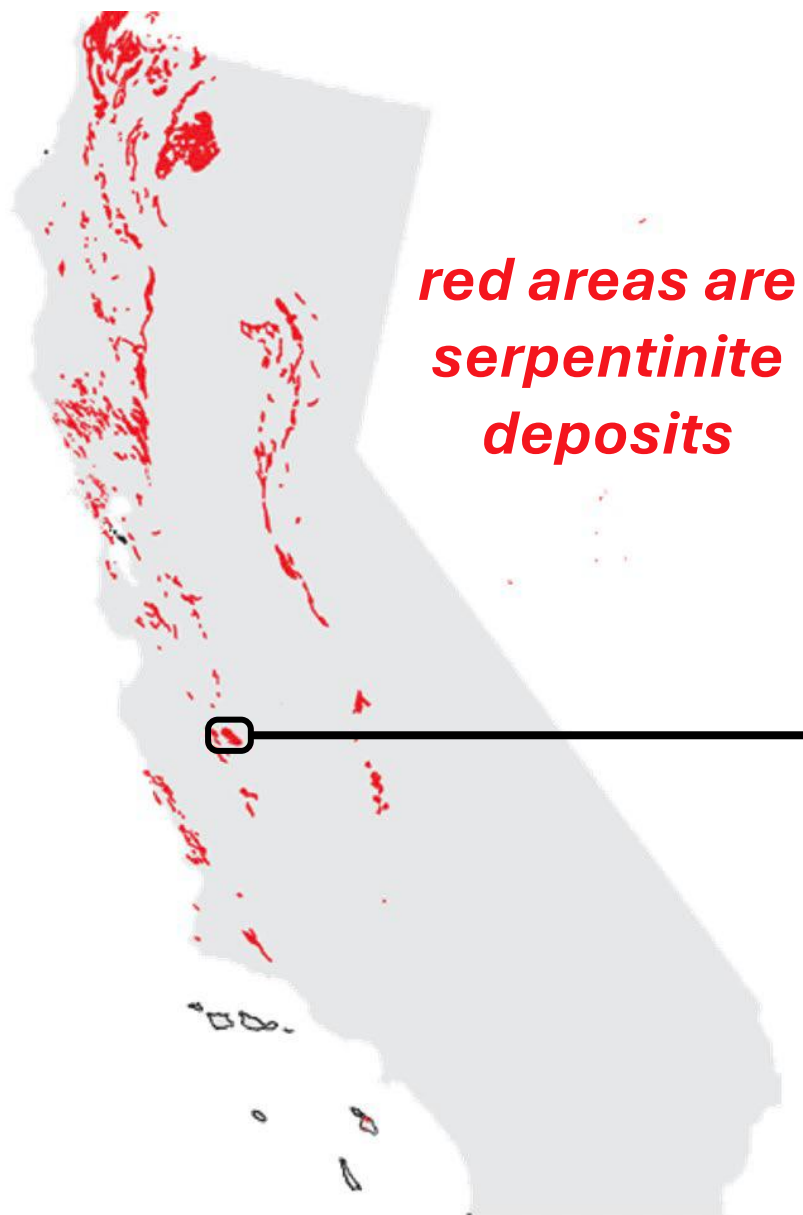


# Q: What's California's Potential?

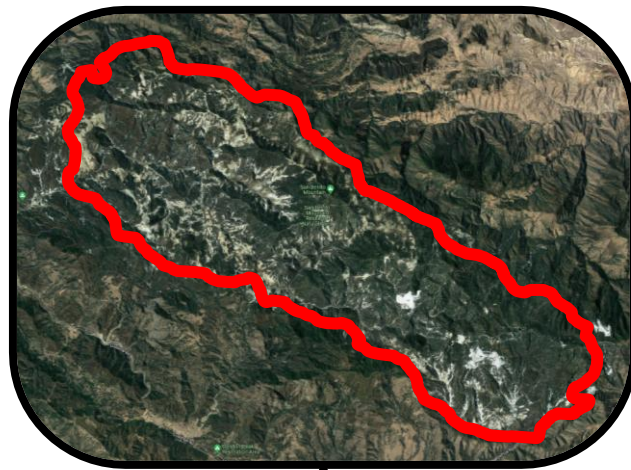




# Q: What's California's Potential?



*red areas are  
serpentine  
deposits*



New Idria Deposit:  
~142 km<sup>2</sup> (~2% of CA resource)  
→ ~50–100 Gt-CO<sub>2</sub> removal  
potential via Anvil's process

Anvil has been working with  
Lawrence Livermore National  
Laboratory via a Strategic Partnership  
Project to develop the New Idria  
deposit for CDR in California.





Anvil Capture Systems

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